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(6th Semester)

PHYSICS

TENTH PAPER

(Nuclear Physics—II)

(Pre-revised)

Full Marks : 75

Time : 3 hours

(PART : A—OBJECTIVE)

(Marks : 25)

The figures in the margin indicate full marks for the questions

SECTION—A

(Marks : 10)

Tick (✓) the correct answer in the brackets provided :

1×10=10

1. Isotones are nuclei with equal number of

- (a) neutrons () (b) protons ()
(c) electrons () (d) shells ()

2. The charge of nucleus is due to the

- (a) electronic configuration () (b) neutrons ()
(c) protons () (d) electrons ()

- 3.** The average life of a radioactive substance and its decay constant are
 (a) directly proportional ()
 (b) reciprocal to each other ()
 (c) inversely proportional ()
 (d) not related ()
- 4.** In radioactive carbon dating, which carbon isotope is used?
 (a) C-14 () (b) C-13 ()
 (c) C-12 () (d) C-11 ()
- 5.** Ultra fast neutron has energy greater than
 (a) 0.5 MeV () (b) 5 MeV ()
 (c) 10 MeV () (d) 20 MeV ()
- 6.** When effective neutron multiplication factor $k < 1$, then the nuclear chain reaction is
 (a) not self-sustaining () (b) self-sustaining ()
 (c) increasing in rate () (d) None of the above ()
- 7.** In linear accelerators, the length of the drift tubes must be proportional to
 (a) $1 : \sqrt{3} : \sqrt{5} : \sqrt{7} \dots$ () (b) $1 : \sqrt{4} : \sqrt{8} : \sqrt{16} \dots$ ()
 (c) $1 : \sqrt{2} : \sqrt{3} : \sqrt{4} \dots$ () (d) $1 : 2 : 3 : 4 \dots$ ()
- 8.** In 1952, Donald A. Glaser of United States invented the
 (a) ionisation chamber () (b) cloud chamber ()
 (c) scintillation chamber () (d) spark chamber ()
- 9.** In cosmic rays, the penetrating ones are called
 (a) hard components () (b) soft components ()
 (c) primary rays () (d) secondary rays ()
- 10.** How many quarks does baryon have?
 (a) 1 () (b) 2 ()
 (c) 3 () (d) 4 ()

SECTION—B

(Marks : 15)

Answer the following questions :

3×5=15

1. Write a note on nuclear charge.
2. What do you mean by half-life of radioactive nuclei?
3. Discuss in brief the origin of stellar energy.
4. Discuss in brief the accelerators in India.
5. Write a note on the concept of anti-particles.

(PART : B—DESCRIPTIVE)

(Marks : 50)

The figures in the margin indicate full marks for the questions

1. (a) What is binding energy? Explain how binding energy affects the stability of a nucleus. 5
(b) Describe packing fraction. Write the relation between packing fraction and mass defect. 5

OR

- (a) Explain the concept of nuclear spin. 2
(b) Discuss semi-empirical mass formula and its simple application. 4
(c) Draw and explain nuclear stability curve. 4
2. (a) Discuss the different types of radioactive decays. 4
(b) What are radioisotopes? Write the production and uses of radioisotopes. 6

OR

- (a) Describe γ -rays and their origin. 4
(b) Explain the three types of β -decay processes. 6

3. (a) Describe nuclear shell model in detail. 5
(b) Discuss the discovery of neutron. Write the properties of neutron. 5

OR

- (a) Write the concept of critical size in nuclear fission reaction. 2
(b) What do you mean by multiplication factor? 2
(c) What is nuclear fusion? Explain how energy is released in fusion reaction. 6

4. Describe the construction, characteristic, working and counting rate of Geiger-Müller (GM) Counter. 10

OR

Explain the following in detail : 5×2=10

- (a) Proportional counter
(b) Cerenkov counter

5. (a) What are primary and secondary cosmic rays? 2
(b) Explain in detail the latitude effect of cosmic rays. 3
(c) Discuss cosmic ray showers and explain the absorption of secondary rays. 5

OR

- (a) What do you understand by leptons? How many leptons are there in total? 4
(b) Describe the following : 2×3=6
(i) Strangeness
(ii) Isospin
(iii) Hypercharge

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